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Reference Tables for  
Physical

Setting/Chemistry -  
2011 Edition 7 Table O

Symbols Used in  
Nuclear Chemistry

Table P Organic  
Prefixes Table Q

Homologous Series of  
Hydrocarbons Name

Notation Symbol alpha  
particle  $4\ 2\ \text{He}$  or  $4$

$2\alpha$  beta particle  $0\ -1$   
e or  $0\ -1\beta$  - gamma

radiation  $0\ 0\gamma$  neutron

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1 0 nn proton 1 1 H or  
1 1 pp positron 0 +1 ...

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question. Question 1:  
0:32 Question 2: 4:18  
Question 3: 7:10  
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Question 7 ...

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Physical  
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E. Selected Polyatomic  
Ions. Formula. Name.  
Formula. Name.  $\text{H}_3\text{O}^+$   
+ hydronium.  $\text{CrO}_4^{2-}$   
42-chromate.  $\text{Hg}_2^{2+}$   
mercury(I)  $\text{Cr}_2\text{O}_7^{2-}$   
72-dichromate.  $\text{NH}_4^+$   
ammonium,  $\text{MnO}_4^-$

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4-permanganate.  $C_2H_4$

$3O_2 - NO_2 - \dots$

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Table F Solubility  
Guidelines for Aqueous  
Solutions 0 102030405  
060708090100  
Temperature (°C)

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## Regents

Solute per 100 g of H<sub>2</sub>O (g)

140	130	120	110
100	90	80	70
60	50	40	30
20	10	0	

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KI NaNO<sub>3</sub>  
NH<sub>3</sub> HCl NaCl KNO<sub>3</sub>  
NH<sub>4</sub>Cl KCl SO<sub>2</sub> KClO<sub>3</sub>

Table G Solubility

Curves Ions That Form  
Soluble Compounds

Exceptions ...

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Physical Setting/  
Chemistry Table H  
Vapor Pressure of Four  
Liquids 200 propanone  
ethanol water Vapor  
Pressure (kPa) 150  
ethanoic acid 101.3  
kPa 100 50 0 0 25 50  
75 100 125  
Temperature (°C)  
Reference Tables for  
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Chemistry 3 Table I

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Heats of Reaction at  
101.3 kPa and 298 K  
Reaction  $\text{CH}_4(\text{g}) +$   
 $2\text{O}_2(\text{g}) \Delta H (\text{kJ}) \dots$

## **Chemistry Regents Reference Table [k6nq27d3j1nw]**

Table S Questions: 1.  
Draw an empty  
periodic table (omit  
groups 3-12) and  
include Symbols,  
Atomic Number, and  
Electronegativity value  
for each element. a.  
Draw a bar graph

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representing the trend  
of electronegativity in  
group 1 and 2.

Organize symbols of  
elements on the X axis  
and the numerical  
values on the Y axis. b.

## **Reference Table**

### **Review - Geneseo**

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Tables A-R. Periodic

Table. Chemistry

Reference Tables S and

T. How to use Table A.

How to use Table B.

How to use Table C.



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